

# EVALUATION OF A GAS-PHASE ATMOSPHERIC MECHANISM FOR LOW NO<sub>x</sub> CONDITIONS

William P. L. Carter

College of Engineering Center for Environmental Research and Technology  
University of California, Riverside, CA 92521

May 27, 2004

- Background and Objectives
- Environmental Chambers Employed
- Chamber Effects Characterization
- Evaluation Results
- Modeling assessment of relative rates of low NO<sub>x</sub> reactions
- Discussion and Conclusions

## BACKGROUND

- Ground-level ozone O<sub>3</sub> is formed in complex reactions of emitted VOCs with NO<sub>x</sub> in sunlight
- The mechanisms representing these reactions in models are critical to accurate O<sub>3</sub> control strategy predictions.
- Environmental chamber data provide the best way to test these mechanisms independent of other uncertainties
- However, current mechanisms were evaluated using data with higher NO<sub>x</sub> levels than most current ambient conditions.
- This is a concern because the nature of the oxidation processes change as NO<sub>x</sub> is reduced.
- Opportunities exist for lower NO<sub>x</sub> mechanism evaluation:
  - Existing Low NO<sub>x</sub> data from CSIRO and TVA Chambers
  - Experiments from the new UCR EPA Chamber designed for low NO<sub>x</sub> evaluations are becoming available

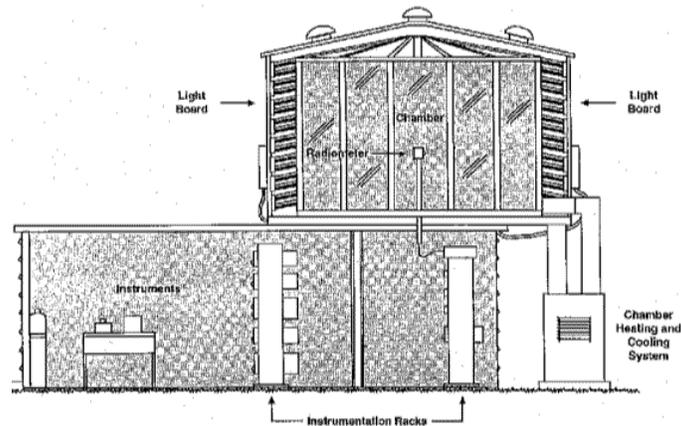
## OBJECTIVES

- Evaluate the SAPRC-99 mechanism (the most up-to-date and detailed mechanism used by the CARB) for low NO<sub>x</sub> conditions.
- Obtain and characterize existing TVA and CSIRO chamber data for mechanism evaluation
- Conduct experiments in the new UCR EPA chamber most needed for low NO<sub>x</sub> evaluation
- Evaluate mechanism using available TVA, CSIRO, and UCR EPA chamber data
- Investigate modifications to SAPRC-99 to better represent low NO<sub>x</sub> conditions
- Recommend research needed to improve mechanism performance

## SUMMARY OF ENVIRONMENTAL CHAMBERS USED IN THE LOW NO<sub>x</sub> EVALUATION

	TVA	CSIRO	UCR EPA
Volume (m <sup>3</sup> )	28	2 x 20	2 x 85
Walls	FEP Teflon® Film		
Lighting	Fluorescents	Sunlight	Argon Arc
Number of Experiments	32 Char. 48 Mech Eval	19 Mixture experiments	36 Char 42 Mech Eval
Dates of Runs	1993-96	1995-1996	2003
Measured Species Used in Evaluation	O <sub>3</sub> , NO, CO, HCHO, PAN, VOCs	O <sub>3</sub> , NO, NO <sub>y</sub> -NO	O <sub>3</sub> , NO, NO <sub>2</sub> , CO, HCHO, PAN, HNO <sub>3</sub> , VOCs

## DIAGRAM OF TVA INDOOR CHAMBER



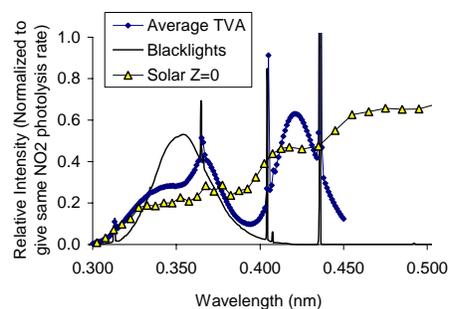
W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

5

## DISCUSSION OF TVA INDOOR CHAMBER

- Experiments conducted by Simonaitis and Bailey of TVA in 1993 through 1995 for low NO<sub>x</sub> mechanism evaluation
- Steps taken to reduce background by extensive purging between runs
- 3 types of fluorescent lamps used to approximate solar spectrum
- NO<sub>2</sub> actinometry results averaged 0.392 min<sup>-1</sup>
- Temperature during runs varied from ~295 - 315 °K
- TVA Chamber no longer operational
- Data made available for modeling by Jeffries and Co-workers for an RRWG project



W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

6

## CSIRO OUTDOOR CHAMBERS



W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

7

## DISCUSSION OF CSIRO OUTDOOR CHAMBER

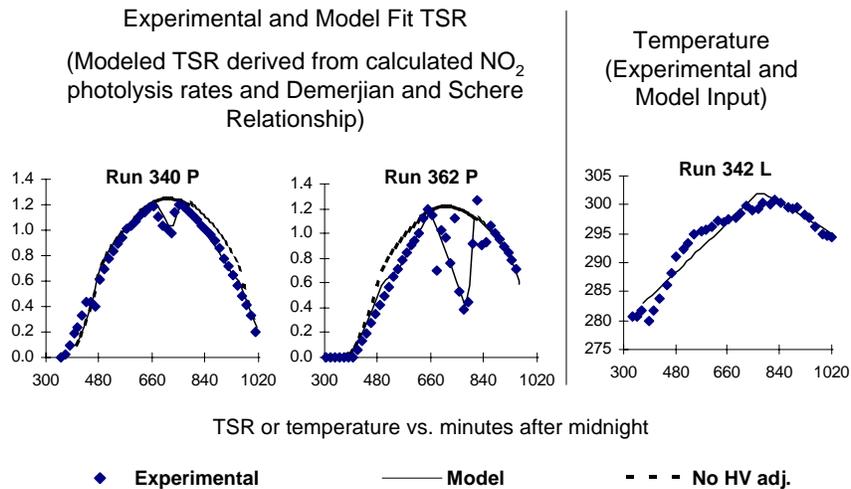
- Located outdoors in a suburb of Sydney, Australia
- Multiple complex surrogate – NO<sub>x</sub> experiments conducted by Johnson and co-workers to test and derive parameters for Johnson's parameterized "extent of reaction" model
- Data used from 10 dual chamber surrogate experiments conducted in a collaborative project with Jeffries and co-workers
- Data made available for modeling as part of an RRWG project
- Photolysis rates for modeling derived as follows:
  - NO<sub>2</sub> photolysis rates as function of time estimated from TSR data using relationship of Demerjian and Schere
  - Ratios of other photolysis rates to NO<sub>2</sub> calculated using Peterson (1977) actinic fluxes as function of time of day
- Other characterization parameters estimated based on characterization results for other chambers

W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

8

## EXAMPLES OF CSIRO TEMPERATURE AND LIGHT CHARACTERIZATION ASSIGNMENTS

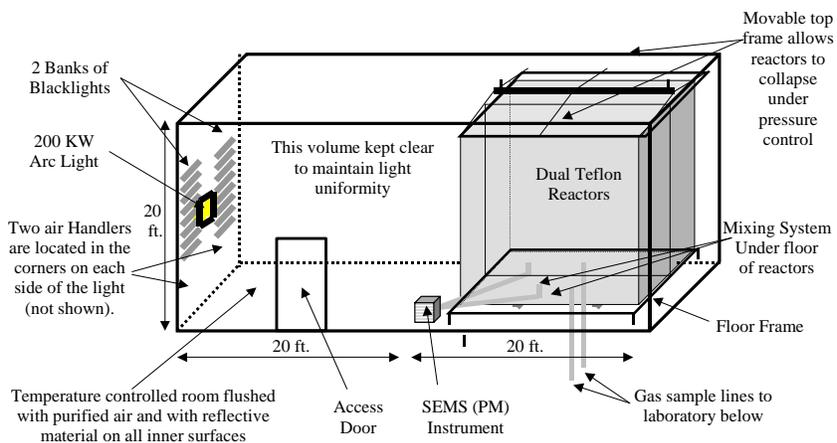


W. P. L. Carter 5/27/04

Low NOx Mechanism Evaluation

9

## DIAGRAM OF UCR EPA CHAMBER

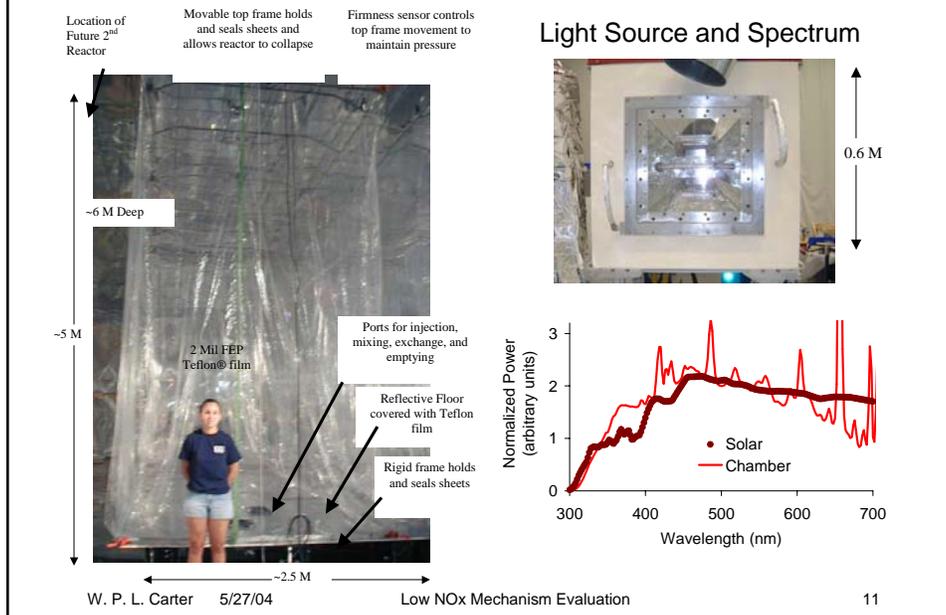


W. P. L. Carter 5/27/04

Low NOx Mechanism Evaluation

10

## PICTURES OF UCR EPA CHAMBER



## DISCUSSION OF UCR EPA CHAMBER

- Constructed using a \$2.9 Million EPA earmark to develop a “Next Generation” chamber for mechanism evaluation
  - Large volume to minimize background and permit PM studies and instrumentation with large sample requirements
  - Indoor chamber for maximum characterization and control
  - Light source simulating sunlight
  - Low background to permit well-characterized experiments at low pollution levels
  - Advanced analytical instrumentation
  - Temperature control to  $\pm 1^\circ\text{C}$  in  $\sim 5^\circ - \sim 50^\circ\text{C}$  range
- Experiments with current configuration began in early 2003:
  - Initial characterization and evaluation runs funded by EPA
  - Very low NO<sub>x</sub> surrogate experiments run for this project

## DERIVATION OF MAJOR CHAMBER CHARACTERIZATION PARAMETERS

Parameter	TVA	CSIRO	UCR EPA
Continuous Radical source (HONO offgasing)	Negligible compared to high HCHO offgasing	T-Dependent rate assumed to be same as SAPRC OTC or 3 x higher	Model char. runs sensitive to HONO offgasing
NO <sub>x</sub> Offgasing	Model CH <sub>3</sub> CHO - Air runs	Assume same as radical source	Approx. same as radical source
HCHO background and offgasing	Model HCHO in CO - NO <sub>x</sub> or NO <sub>x</sub> - air runs	Assumed to be unimportant compared to R.S.	Model HCHO in char. runs w/o HCHO source
Initial HONO	Model CO - NO <sub>x</sub> , NO <sub>x</sub> - Air, and CH <sub>4</sub> - NO <sub>x</sub> runs	No data – varied in evaluation simulations	Model char. runs sensitive to initial HONO
O3 Decay	Measured	Estimated	Measured
Dilution	Derived from tracer data	Estimated to be minor	Believed to be low in most runs

W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

13

## VALUES OF MAJOR CHAMBER CHARACTERIZATION PARAMETERS

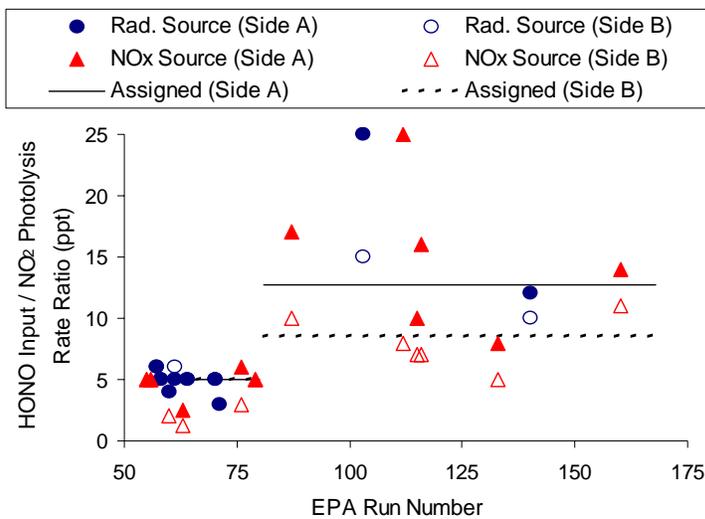
Parameter	TVA	CSIRO (varied)	UCR EPA
HONO or NO <sub>x</sub> offgasing / NO <sub>2</sub> photolysis rate ratio	7.2 ppt (as NO <sub>x</sub> offgasing only)	<u>Standard</u> : Temp.-dependent fit for OTC (10-100 ppt) <u>High RS</u> : 3 x Std.	<u>Runs 55-80</u> : 5 ppt <u>Runs 81-168</u> : •Side A: 8.5 ppt •Side B: 12.5 ppt
HCHO offgasing / NO <sub>2</sub> photolysis rate ratio	<u>HCHO</u> : 45 ppt <u>HCHO precursor</u> : 135 ppt	Assumed to be low compared to other sources	10 ppt
Initial HONO	0.5 ppb	<u>Standard</u> : ~0 <u>High RS</u> : 2 ppb	0.05 ppb
O3 Decay	7% /hour	1.3% /hour	1.1% /hour
Light Intensity	k <sub>1</sub> = 0.392 min <sup>-1</sup>	<u>Standard</u> : Deriv'd from TSR data <u>High</u> : 1.15 x Std.	k <sub>1</sub> = 0.284 min <sup>-1</sup>

W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

14

### UCR-EPA HONO INPUT PARAMETER VS. RUN NO.

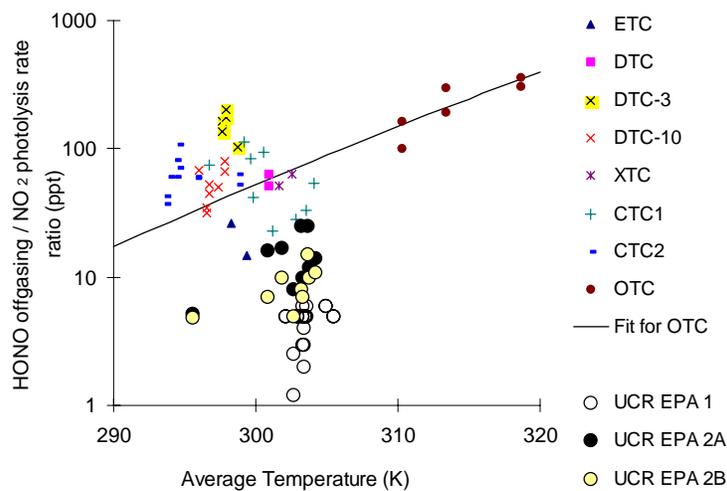


W. P. L. Carter 5/27/04

Low NOx Mechanism Evaluation

15

### COMPARISONS OF RADICAL SOURCE PARAMETERS FOR UCR EPA AND PREVIOUS CHAMBERS



W. P. L. Carter 5/27/04

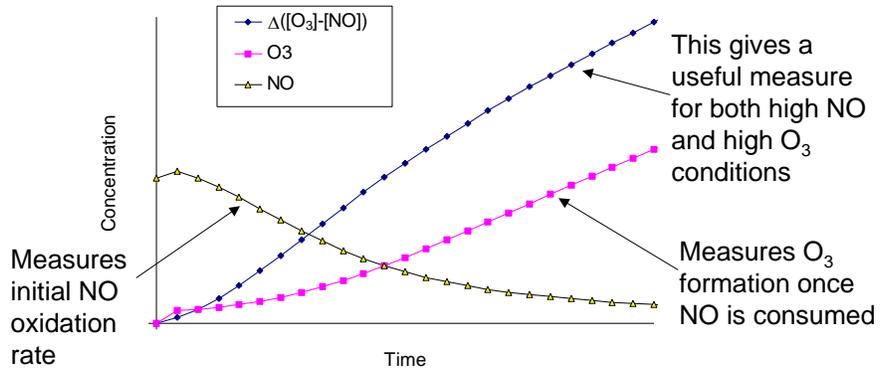
Low NOx Mechanism Evaluation

16

## MEASURE OF MODEL PERFORMANCE FOR OZONE

Model performance for simulating both ozone formation and NO oxidation is measured by ability to predict  $\Delta([\text{O}_3]-[\text{NO}])$ :

$$\Delta([\text{O}_3]-[\text{NO}]) = ([\text{O}_3]-[\text{NO}])_{\text{FINAL}} - ([\text{O}_3]-[\text{NO}])_{\text{INITIAL}}$$



W. P. L. Carter 5/27/04

Low NOx Mechanism Evaluation

17

## UCR EPA RUNS AND FITS TO $\Delta([\text{O}_3]-[\text{NO}])$

Run Type	Runs	NO <sub>x</sub> Range (ppb)	Average Model Fits	
			Bias	Error
Characterization	32	0 - 200	-2%	31%
Formaldehyde - NO <sub>x</sub>	2	8 - 25	-23%	23%
Formaldehyde - CO - NO <sub>x</sub>	2	15-20	-10%	10%
Ethene - NO <sub>x</sub>	2	10 - 25	-15%	15%
Propene - NO <sub>x</sub>	2	5 - 25	16%	16%
Toluene or m-Xylene - NO <sub>x</sub>	4	5 - 25	10%	10%
Aromatic - NO <sub>x</sub> + CO	6	5 - 30	-17%	18%
Ambient Surrogate - NO <sub>x</sub>	24	2 - 110	-11%	15%

W. P. L. Carter 5/27/04

Low NOx Mechanism Evaluation

18

### SUMMARY OF TVA RUNS AND FITS TO $\Delta([O_3]-[NO])$

Run Type	Runs	NO <sub>x</sub> Range (ppb)	Average Model Fits	
			Bias	Error
Characterization	32	0 - 54	1%	15%
Formaldehyde - NO <sub>x</sub>	4	39 - 42	-4%	10%
Isopentane - NO <sub>x</sub>	1	18	-28%	28%
Ethene, Propene, or trans-2-Butene - NO <sub>x</sub>	7	22 - 54	10%	10%
Toluene or m-Xylene - NO <sub>x</sub>	5	50 - 266	1%	8%
Simple Mixture - NO <sub>x</sub>	23	50 - 100	6%	7%
Ambient Surrogate - NO <sub>x</sub>	12	25 - 169	-8%	9%

W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

19

### CSIRO RUNS AND FITS TO $\Delta([O_3]-[NO])$

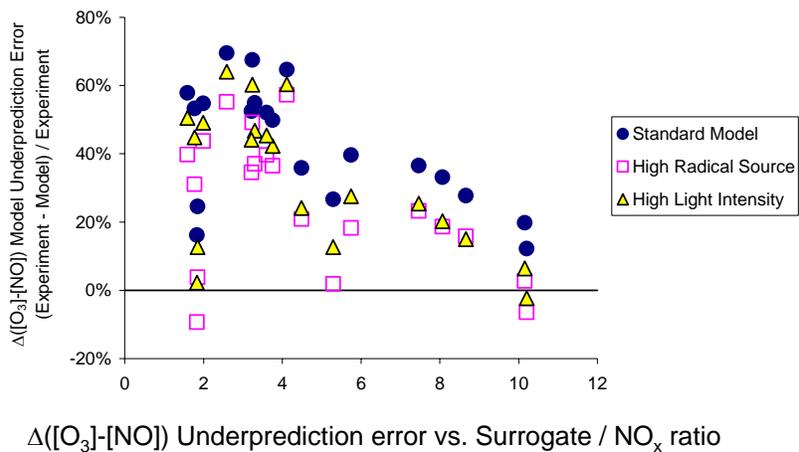
Run Type	Runs	NO <sub>x</sub> Range (ppb)	Average Model Fits	
			Bias	Error
Ambient Surrogate – NO <sub>x</sub>	20	17 - 100		
Standard Char. Model			-42%	42%
High Radical Source			-26%	27%
High Light Intensity			-33%	33%

W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

20

## EFFECTS OF ALTERNATIVE CHARACTERIZATION ASSUMPTIONS ON CSIRO SIMULATIONS

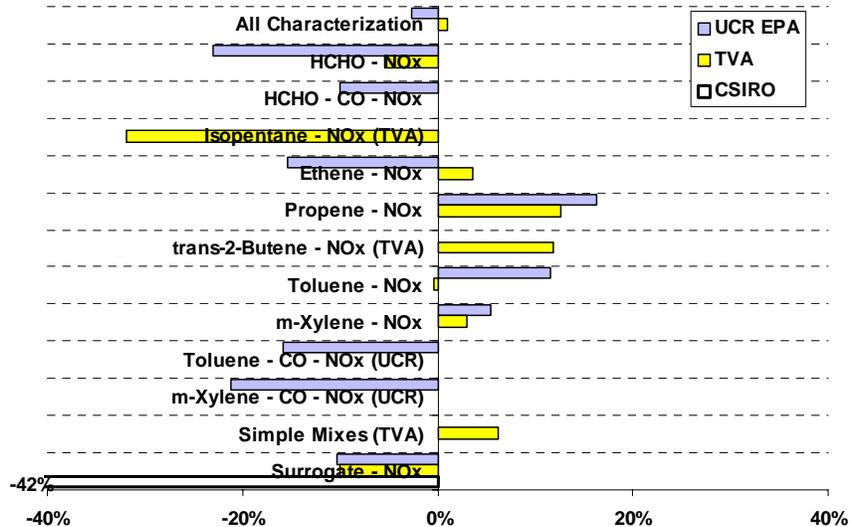


W. P. L. Carter 5/27/04

Low  $NO_x$  Mechanism Evaluation

21

## MODEL BIASES FOR PREDICTION OF $\Delta([O_3]-[NO])$

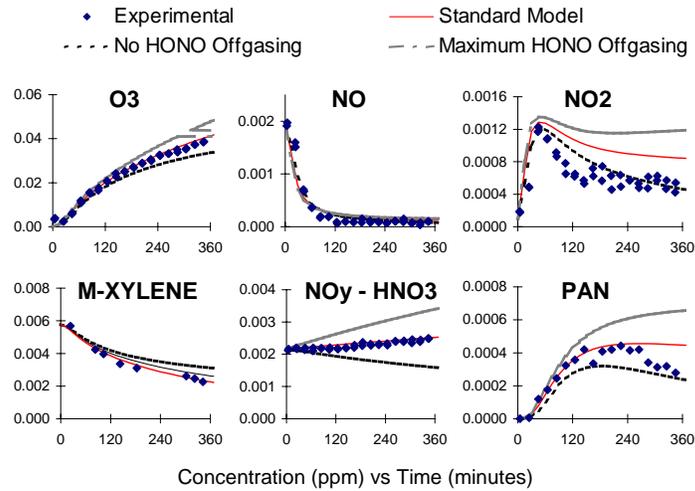


W. P. L. Carter 5/27/04

Low  $NO_x$  Mechanism Evaluation

22

## LOWEST NO<sub>x</sub> UCR EPA SURROGATE EXPERIMENT (ROG SURROGATE = 300 PPBC, NO<sub>x</sub> = 2 PPB)

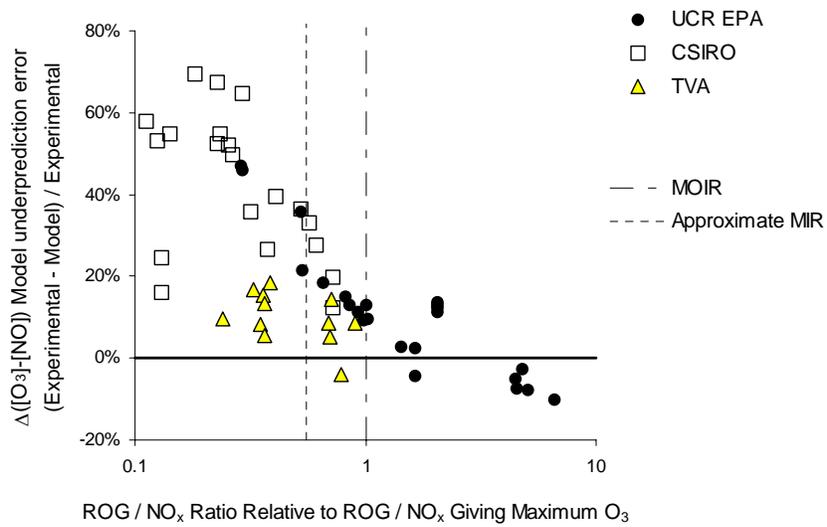


W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

23

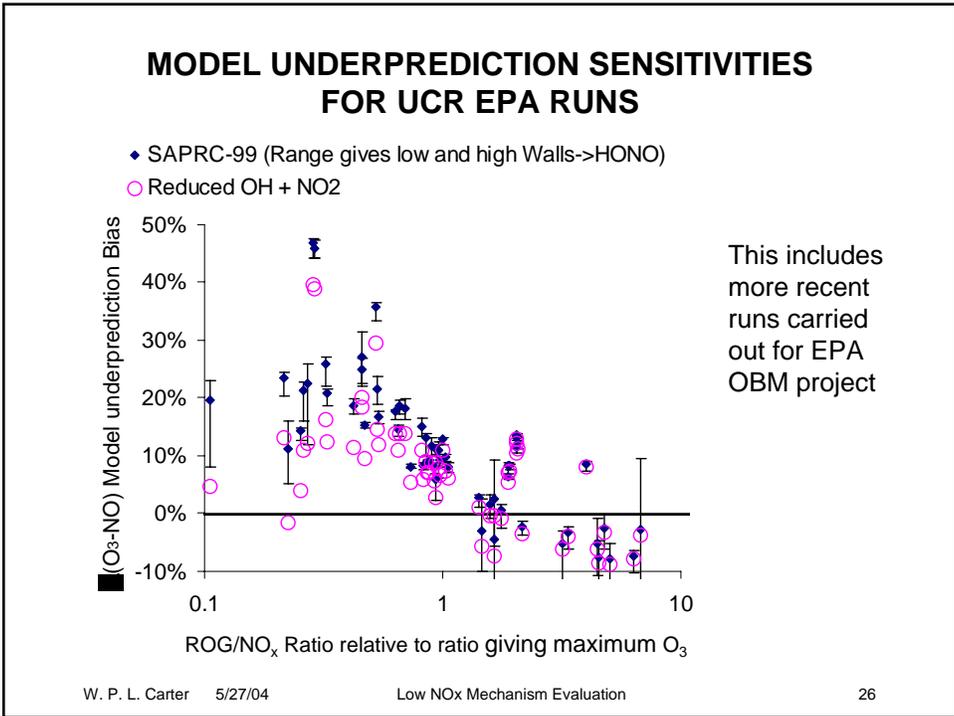
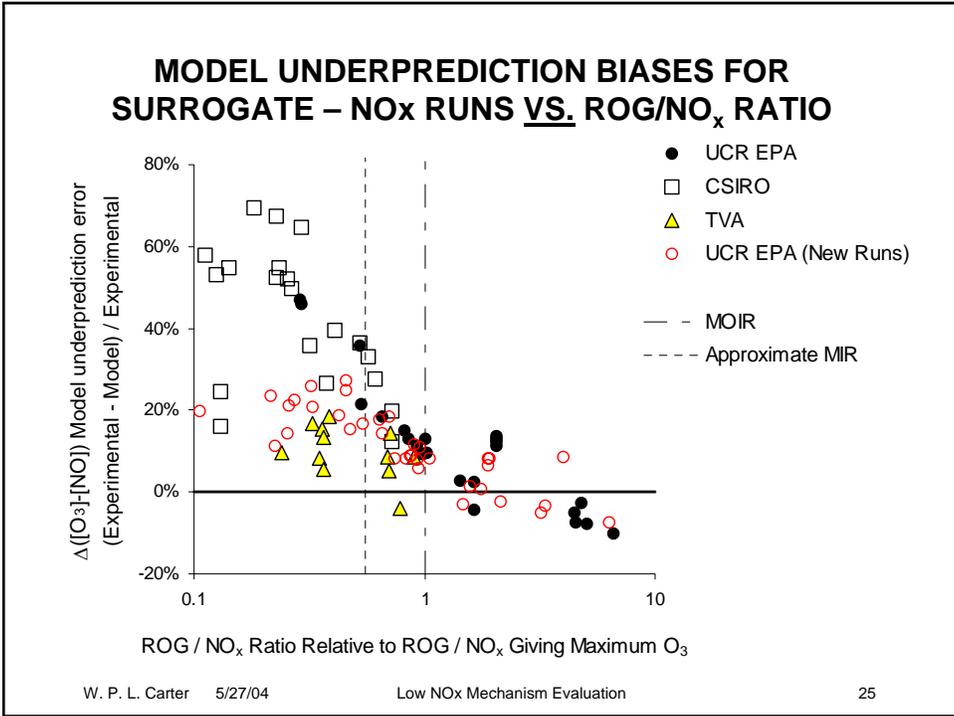
## MODEL UNDERPREDICTION BIASES FOR SURROGATE – NO<sub>x</sub> RUNS VS. ROG/NO<sub>x</sub> RATIO



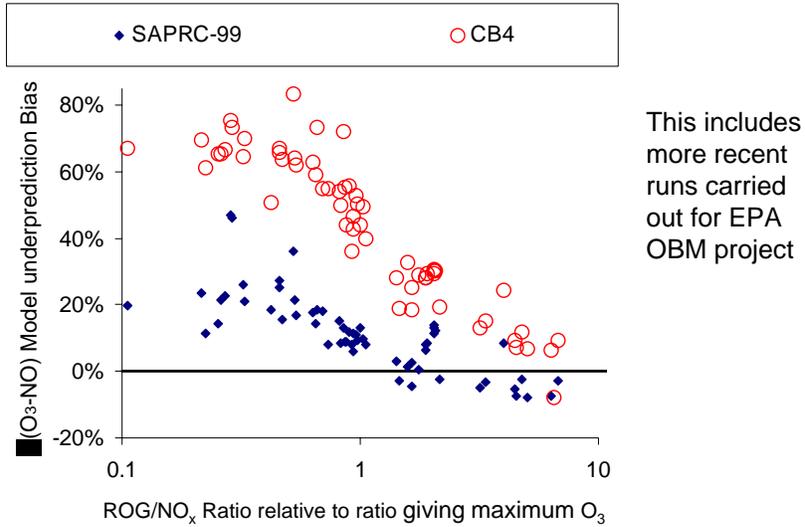
W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

24



## COMPARISON OF CB4 AND SAPRC-99 MODEL ERRORS FOR UCR EPA SURROGATE RUNS

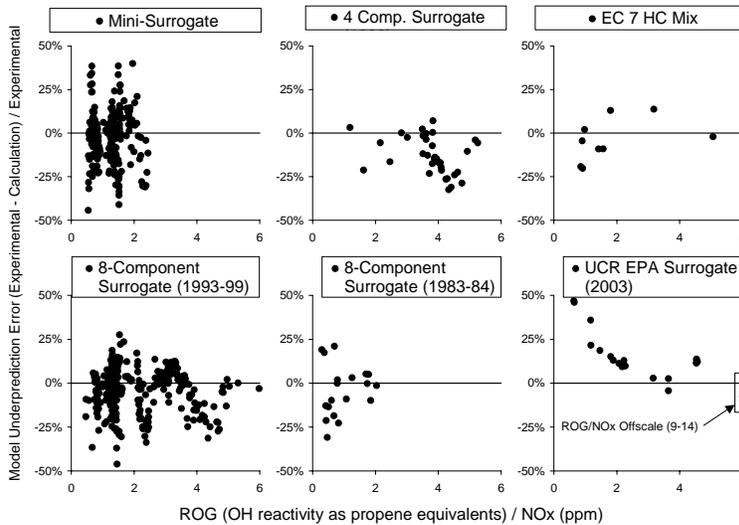


W. P. L. Carter 5/27/04

Low NOx Mechanism Evaluation

27

## COMPARISON OF CURRENT SURROGATE FITS WITH DATA USED IN PREVIOUS EVALUATION



W. P. L. Carter 5/27/04

Low NOx Mechanism Evaluation

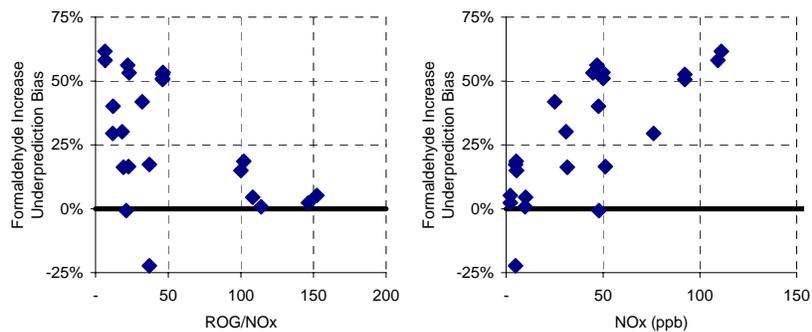
28

## MODEL PERFORMANCE IN SIMULATING OTHER MEASUREMENTS IN SURROGATE RUNS

Compounds	TVA	UCR EPA
NO <sub>2</sub>	No data (NO <sub>y</sub> -NO only)	Tends to underpredict consumption following maximum
PAN	Very significantly underpredicted	Reasonably well simulated in runs where O <sub>3</sub> well simulated
Formaldehyde	Tendency to underpredict (depends on offgasing model)	Tendency to underpredict depends on NO <sub>x</sub> levels
Reactant VOCs	Generally consistent with data	Tendency to underpredict final consumption rates
HNO <sub>3</sub>	No data	Generally consistent with available data ([NO <sub>x</sub> ] <sub>i</sub> >50 ppb)

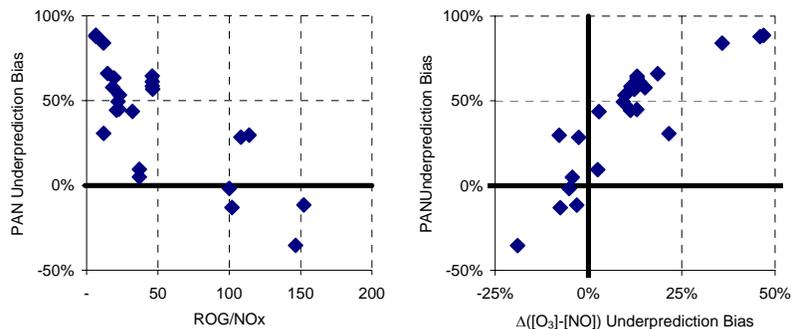
Note: Data for above compounds not available for CSIRO Runs

## MODEL PERFORMANCE SIMULATING FORMALDEHYDE IN UCR EPA SURROGATE RUNS



Formaldehyde Underprediction bias vs. ROG/NO<sub>x</sub> or NO<sub>x</sub>  
(experimental - calculated) / experimental Δ[HCHO]

## MODEL PERFORMANCE SIMULATING PAN IN UCR EPA SURROGATE RUNS



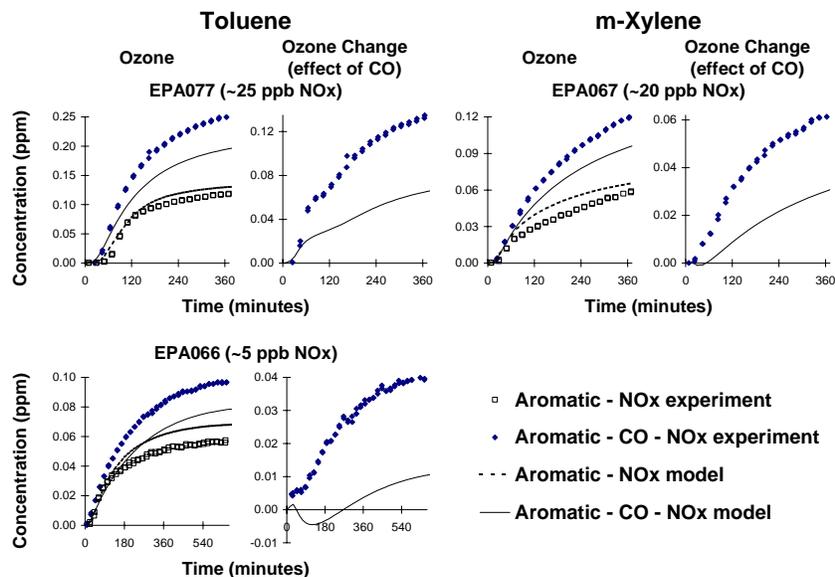
PAN Underprediction bias vs. ROG/NO<sub>x</sub> or NO<sub>x</sub>  
(experimental - calculated) / experimental final PAN

W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

31

## EFFECT OF CO ON AROMATIC - NO<sub>x</sub> RUNS

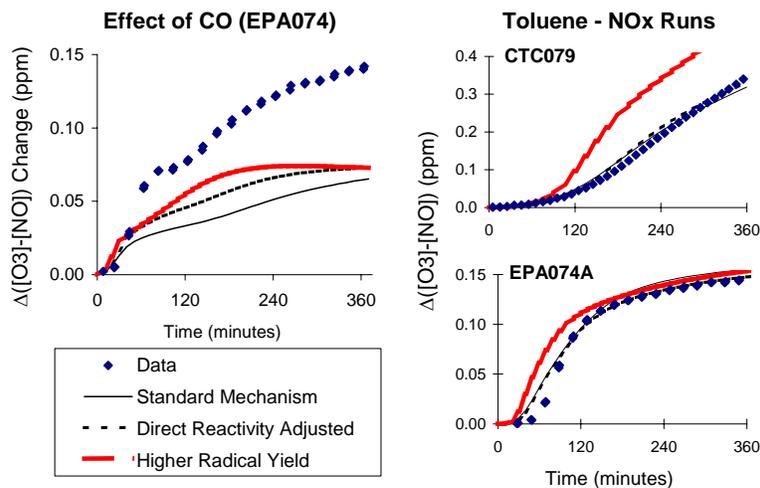


W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

32

## EFFECTS OF ADJUSTMENTS TO PARAMETERIZED TOLUENE MECHANISM



W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

33

## ASSESSMENT OF NEED TO MODIFY MECHANISM FOR LOW NO<sub>x</sub> CONDITIONS

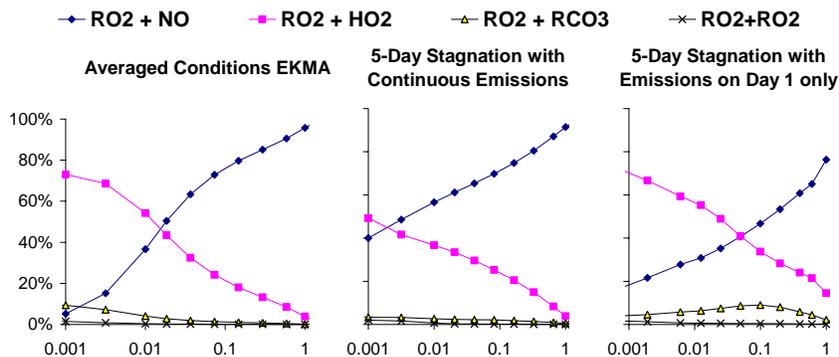
- Lower NO<sub>x</sub> levels result in increased importance of peroxy + peroxy reactions, which form different organic products than the peroxy + NO reactions that dominate when NO<sub>x</sub> is higher.
- SAPRC-99 uses an approximate “chemical operator” method for RO<sub>2</sub> reactions that neglects this change in products with NO<sub>x</sub>
- Representing RO<sub>2</sub>+RO<sub>2</sub> reactions more explicitly requires adding many reactions to the mechanism.
- Process analysis calculations were carried out to assess the relative importance of the different types of competing RO<sub>2</sub> reactions at low NO<sub>x</sub>
- The results can then be used to assess priorities for mechanism modifications for more accurate low NO<sub>x</sub> predictions

W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

34

## MODEL SIMULATIONS OF PEROXY REACTION INTEGRATED REACTION RATE RATIOS



Ratios of integrated reaction rates vs.  
 NO<sub>x</sub> inputs relative to NO<sub>x</sub> giving maximum O<sub>3</sub>

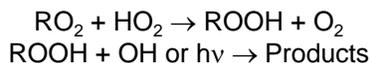
W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

35

## IMPLICATIONS CONCERNING HOW TO MODIFY MECHANISMS FOR LOW NO<sub>x</sub> CONDITIONS

- Results of test calculations of integrated peroxy radical reaction rates with three types of low NO<sub>x</sub> scenarios indicate that:
  - Major low-NO<sub>x</sub> sink for peroxy radicals is reaction with HO<sub>2</sub>
  - Maximum importance of Alkyl Peroxy + Acyl Peroxy reaction is ~10%
  - Alkyl peroxy + Alkyl peroxy reactions negligible
- Adding reactions or species to improve representations of organic peroxy + peroxy reactions probably not worthwhile
- Higher priority is improving representation of:



- This requires adding more hydroperoxide species to the mechanism, whose reactions are uncertain.

W. P. L. Carter 5/27/04

Low NO<sub>x</sub> Mechanism Evaluation

36

## DISCUSSION AND CONCLUSIONS:

### Good News

- Range of conduction where mechanisms have been evaluated has been significantly expanded
  - TVA data of comparable or better quality than previous runs in other chambers, despite formaldehyde contamination
  - CSIRO data proved to be useful in this evaluation, but more characterization information would increase its utility
  - Lower background in UCR EPA chamber permitted useful mechanism evaluation with NO<sub>x</sub> as low as ~2 ppb and improves precision of evaluation
  - Simulations of lower NO<sub>x</sub> characterization and most simple VOC – NO<sub>x</sub> experiments generally satisfactory
- No apparent mechanism problem simulating very low NO<sub>x</sub> conditions where maximum O<sub>3</sub> formation potentials achieved.
- Inaccuracies caused by approximate treatment of RO<sub>2</sub>+RO<sub>2</sub> reactions in current mechanism are probably not important

## DISCUSSION AND CONCLUSIONS:

### Bad News

- SAPRC-99 mechanism has consistent bias in underpredicting NO oxidation and O<sub>3</sub> formation rates at low ROG/NO<sub>x</sub> ratios
- Underprediction bias for CB4 even worse
- Significant problems with current aromatics mechanisms:
  - Ozone increase caused by adding CO “radical amplifier” to aromatic – NO<sub>x</sub> systems underpredicted by factor of ~2
  - Adjustments to the mechanisms as currently parameterized cannot correct this problem
  - Aromatics mechanism problems *may* be the cause of the low ROG/NO<sub>x</sub> underprediction bias, but this is not certain.
- Improving accuracy of low NO<sub>x</sub> organic product predictions for requires explicit treatment of uncertain organic hydroxides

## RESEARCH NEEDS

- The current parameterized aromatics mechanisms need to be reformulated and made consistent with available data
- The underprediction problem at low ROG/NO<sub>x</sub> needs to be investigated and resolved
- Data are needed on effects of temperature on ozone and other secondary products. (UCR EPA chamber suitable for this)
- Well-characterized, low NO<sub>x</sub> chamber data are needed to develop and test models for secondary PM formation. (UCR EPA chamber also suitable for this)
- SAPRC-99 needs to be updated to be consistent with latest recommendations and made more compatible with SOA models
- A new condensed mechanism, traceable to an updated and evaluated detailed version, needs to be developed to finally replace the out-of-date CB4 mechanism